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ART. VI.—Observations on the Mineral and Vitamin Content of Australian Milk.

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[From the Australian Institute of Anatomy, Canberra.] [Read 13th July, 1939; issued separately, 1st March, 1940.]

### Introduction.

The vitamin content of cow's milk has been shown to vary within reasonably wide limits, while it is well known that the mineral content, although remarkably constant, also varies within what might be termed physiological limits. During the routine analysis of a number of sweetened and unsweetened condensed milk samples from various parts of the world, Professor J. L. Rosedale, of the King Edward VII. College of Medicine in Singapore, found that the Australian samples were low in vitamin A. In a private communication he suggested that an explanation of the difference may be found in the reputedly low phosphorus content of Australian soils, and some support was given to this hypothesis by Richmond's (1930) suggestion that lecithin, because of its phosphorus content, may be an index of the vitamin (presumably vitamin A) content of butter-fat. Initially, the object of this investigation was to investigate Professor Rosedale's hypothesis and in so doing to determine the vitamin A content of a large number of Australian milk samples. Later, however, it was decided to extend the investigation with the object of discovering any correlations which might exist between the major vitamin and mineral constituents of milk.

One hundred and sixty-eight samples of mixed milk, representing the milk from 6,460 cows, were collected over a period of twelve months from New South Wales, Victoria and Tasmania. One hundred and twenty-five samples were known The mineral constituents to be from afternoon milkings. estimated were :--- calcium, phosphorus, potassium, sodium, each sample within twelve hours of collection and the milk was only exposed to direct daylight at the time of collection and while the estimation was being carried out in the laboratory. The methods employed for the estimation of the vitamin constituents were chosen because they involved the minimum amount of time and gave results, the relative values of which were reliable. But because the methods employed were non-biological the results are considered comparative only, although during preliminary work results obtained by these methods were checked against results obtained by well-controlled biological methods for the same samples and, so far as the results could be compared, they were in good agreement.

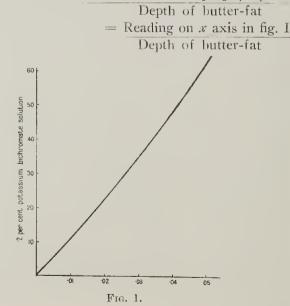
#### Methods of Analysis.

Fat was estimated by Richmond's (1930) modification of the Roese Gottlieb method. The specific gravity was determined with a Westphal balance and corrected for temperature.

VITAMIN A AND CAROTENE.

A pint of milk was centrifuged, the skim milk separated and kept for later determinations, the cream churned into butter. portion of the butter dissolved in petroleum ether and the whole transferred to a separating funnel containing distilled water. The ethereal layer was separated, washed and evaporated under suction at about 30°C. Because facilities were not available for estimating carotene spectro-photometrically, it was estimated in the warm residual butter-fat by a modification of Palmer's (1922) colorimetric method. Several workers have found that for butter-fat the graph given by Palmer gives carotene values which are several times too high, and Barnett (1934) has obtained a correction factor .28 which enables more accurate results to be obtained. However, Barnett assumes that carotene is the only colouring matter of consequence in butter-fat, but Gillam (1934) has shown that the ratio of carotene to xanthophyll in English butter-fat is fairly constant, being approximately 14:1 by weight, and has estimated the carotene value at approximately 94 per cent. of the total yellow colour of the butter-fat. This result is supported by the work of Baumann and Steenbock (1933) on an American butter-fat. Hence a more accurate conversion factor would be .263. Introducing this modification and simplifying Palmer's expression we have :-

Percentage carotene ==  $.00268 \times .263 \times K_2 Cr_2 O_7$  equivalent



To estimate vitamin A, 2 gms. of butter-fat were saponified with 15 cc. of N/2 alcoholic potassium hydroxide, 20 ccs. of water added, the whole transferred to a separating funnel and extracted with two quantities of 25 ccs. of ether. The ethereal extracts were washed with water, then with N/2 aqueous potassium hydroxide and again with water, after which the ether was evaporated under suction at about 30°C. The residue was dissolved in purified ethyl alcohol, made up to 10 ccs. and vitamin A estimated in the solution with a Hilger Vitameter A using the factor 1600.

#### VITAMIN $B_1$ .

In 1935, Schopfer published a paper showing that vitamin  $B_1$  is a growth factor for the mould *Phycomyces blakesleeanus*, and he outlined a method for the estimation of vitamin  $B_1$ . It was later found (Robbins and Kavanagh (1937), Sinclair (1937), Schopfer and Jung (1937)) that both vitamin  $B_1$  and its degradation products promote the growth of the mould. Hence the method about to be described for the estimation of vitamin  $B_1$  actually estimates vitamin  $B_1$  and any of its breakdown products which might be present in fresh milk.

Into 50 cc. Erlenmeyer flasks were placed 0.2 cc. of skim milk, each milk sample to be analysed, being done in triplicate. Skim milk was used, for otherwise a thin layer of butter-fat settled on the surface of the media and the slight anaerobic conditions thus introduced inhibited the growth of the mould. To each flask was then added 10 ccs. of a medium consisting of :--

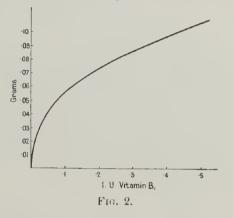
Glucose				166.8	gms.
Asparagin				6.4	"
$MgSO_4 +$	$7\mathrm{H_{2}O}$	• •	· •	0.84	2.9
$\mathrm{KH}_{2}\mathrm{PO}_{4}$			• •	2.5	2.2
$H_2O$	• •	• •	• •	1,665	CCS.

and the pH was adjusted to approximately 6.6 by the addition of one or two drops of dilute sodium hydroxide.

A standard range was also set up containing 10 ccs. of media as before, but in place of milk the following amounts of vitamin  $B_1$  were added:—0.5, 0.4, 0.3, 0.2, 0.1, 0.01, 0.001 international units. These flasks were set up in duplicate and all the flasks then plugged and sterilized at 107°C. for 10 minutes. A culture of *Phycomyces blakesleeanus* was prepared a fortnight previously in 100 ccs. of Wort Agar, made as follows:—25 ccs. of malt extract, 4 gms. of agar, and 200 ccs. of water were made up and autoclaved at 110°C. for 15 minutes. Several grams of the spore bearing mycelium were removed with sterile forceps and thoroughly washed in sterile water contained in a beaker covered with a watch glass. Each flask was then inoculated with two drops (about 0.2 cc.) of the spore bearing suspension

by means of a sterile pipette. The suspension was stirred frequently to ensure that the spores did not sediment. The inoculated flasks were then left in the dark at room temperature (22°C.) for ten days. All the mycelia were then removed, washed in running water, alcohol, and ether, rolled into small balls, dried in a hot air oven at 110°C. and then weighed.

From the weights of the mycelia in the standard flasks, a graph, of which Fig. II. is typical, was drawn, and from this graph the amounts of vitamin  $B_1$  in the other flasks calculated.



#### VITAMIN B<sub>2</sub>.

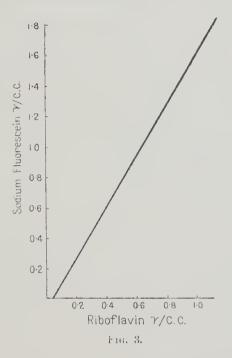
Was estimated in 100 ccs. of skim milk according to the second method described by Weisberg and Levin (1937). The aqueous solution of riboflavin contained much foreign matter, but this did not interfere with the estimation. The relation between the fluorescence intensity of sodium fluorescein (May and Baker Limited) and vitamin  $B_2$  is given in Fig. III., which differs slightly from that given by Weisberg and Levin, for it is a straight line which does not pass through the origin.

#### VITAMIN C.

To 10 ccs. of milk were added 10 ccs. of 20 per cent. trichloracetic acid; the solution was mixed well and then filtered. The precipitate was washed once with a little distilled water and the filtrate made up to 25 ccs. A burette was charged with this solution and it was added drop by drop to a standardized solution of 2:6 dichlorphenolindophenol until the red colour was just discharged. Knowing the strength of the dichlorphenolindophenol solution, the vitamin C present was determined.

#### ASHING.

300 ccs. of milk in a porcelain basin were dried on a water bath, a little alcohol being added to facilitate evaporation. When dry, the residue was heated over a bunsen to set the carbon and then placed in a muffle at a temperature of 400°-450°C. To



complete the incineration, the carbonized mass was cooled, mixed with water, dried, and returned to the muffle. The following determinations were carried out in duplicate on the ash which was dissolved in the minimum amount of hydrochloric acid and the volume made up to 100 ccs. :—

1. Calcium.—25 ces. of the ash solution were made alkaline with ammonia and the resulting precipitate filtered off and washed. The filtrate and washings were kept for the sulphur determination. The precipitate was then dissolved in the minimum amount of hydrochloric acid, sodium acetate solution added and adjustment made to pH 5.0, using congo red as indicator. Calcium was then precipitated at 40°C. with an excess of saturated ammonium oxalate, the solution allowed to stand one hour, filtered, washed with water and a little dilute ammonia, incinerated and weighed as CaO.

2. Magnesium.—The combined filtrate and washings were evaporated to about 100 ccs., and 20 ccs. of nitric acid added. The beaker was covered, evaporated to dryness, 5 ccs, of hydrochloric acid were added, then evaporated almost to dryness and the residue dissolved in hot water. The magnesium was precipitated by the addition of 3 ccs, of a 10 per cent, solution of annonium phosphate and sufficient annonium hydroxide to make the solution slightly alkaline. The solution was stirred vigorously, allowed to stand 15 minutes, 15 ccs, of annonium hydroxide added and the precipitation allowed to proceed over-10856/39.-8

night. It was then filtered, washed with dilute ammonia, transferred to a crucible, dried, ignited, and weighed as  $Mg_2P_2O_7$ .

3. Sulphur.—The filtrate from the calcium estimation to which 10 ccs. of concentrated hydrochloric acid had been added was evaporated almost to dryness. It was then made up to 250 ccs., 5 ccs. of dilute hydrochloric acid added, and boiled. Some acidified 3 per cent, barium chloride was also boiled and 8 ccs. slowly added to the sulphate solution. It was left for an hour, filtered, washed, ignited, cooled, three drops of a mixture containing 1 cc. of alcohol and 2 ccs. of sulphuric acid added, incinerated and weighed as BaSO<sub>4</sub>.

4. Potassium.—To the filtrate from the sulphate determination. barium hydroxide solution was added in slight excess and it was evaporated to about 50 ccs. Ammonium carbonate was added, the solution filtered into a porcelain dish, washed with hot water and evaporated to dryness. Moisture and ammonium chloride were driven off, 5 ccs. of hot water added, the solution filtered into a tared beaker, the residue washed, one drop of hydrochloric acid added, the filtrate evaporated to dryness and weighed as NaCI + KCI. 15 ccs. of 20 per cent. perchloric acid were added and it was evaporated almost to drvness. To the almost dry, cold residue 10 ccs. of a washing solution consisting of 1 cc. of 20 per cent. perchloric acid in 100 ccs. of 98 per cent. alcohol were added. The solution was stirred well, set aside for 5 minutes, decanted through a tared sintered glass crucible, placed on the steam bath to remove alcohol, dissolved in the minimum amount of hot water, evaporated, 10 ccs, of washing solution added, stirred, filtered, dried for one hour at 130°C. and weighed as KClO<sub>4</sub>.

5. Sodium was calculated from the two above results by difference.

6. *Phosphorus.*—The original solution was again made up to 100 ccs. and to 15 ccs. a little methyl orange was added and the solution boiled to drive off carbon dioxide. The solution was neutralized with sodium hydroxide, 25 ccs. of 10 per cent. calcium chloride and a few drops of phenolphthalein added and the solution titrated with N/10 sodium hydroxide until a slight pink remained permanent after mixing. The percentage phosphorus was given by—

# cc. N/10 NaOH $\times$ .1551 weight of milk taken.

7. Iron.—To 35 ccs. of the remaining ash solution diluted to 100 ccs. were added 10 ccs. of concentrated hydrochloric acid. The iron was precipitated with 4 per cent. cupferron, filtered under suction, washed with water, dilute annonia and again with water. The precipitate was transferred to an annealing cup, heated over a bunsen, incinerated and weighed directly as  $Fe_aO_{32}$ .

### Results.

The results of the analyses are given in Tables I. and II.

								· · · · · · · · · · · · · · · · · · ·	
			Per-	Per-	Per-	Per-	Per-	Per-	Per-
Num	aber.	Specific	centage	centage	centage	centage	centage	centage	centage
T. ( 111	LOCI .	Gravity.	Ca.	P.	K.	Na.	Mg.	S.	Fe.
-		1.0015	100	.000	. 150	.044	1011	.000	.000000
$\frac{1}{2}$	• •	1.0315 1.0320	•130 •125	*082 *088	•156 •148	*044 *042	*011 *000	*00'}	*000090 *000108
3	• •	1.0310	•120	•086	•148	.042	.012	•00)	·000108
4		1.0332	· 107	•087	·134	•048	•011	+003	•000081
5		1.0331	•134	+090	·145	•051	•009	+013	·000126
6		1.0330	•118	+097	•148	•050	•013	•003	·000081
7		1.0321	·122	+094	•142	•049	•014	+003	
8		1.0325	·111	•0.00	·138	•051	•011	+007	
9	• •	1.0320	•120	•086	.140	•057	•010	*003	•000108
10	• •	$1 \cdot 0317$ $1 \cdot 0315$	•118	•075 •091	•145 •152	•054 •045	•013	*003	+000108 +000065
$\frac{11}{12}$	• •	1.0317	·125 ·128	•086	150	•010	*014 *012	•010	•000073
13		1.0310	·127	+078	•146	•042	•011	•011	+000068
14		1.0315	•140	•084	·150	•045	•013	*016	•020084
15		1.0311	·119	•080	·144	*047	•012	+003	·000104
16		1.0320	•125	•085	•147	*050	+011	+003	·000070
17		1.0311	•130	•084	•143	°045	.012	•010	+000/181
18	• •	1:0310	•135	*087	•150	•050	*011	*012	·000210
$\frac{19}{20}$	• •	1.0311 1.0312	•116 •125	•079 •087	•135 •142	·058	*009	+009 +009	·000089
20 21		1.0312	•125	*087	*142 •140	*053 *048	*012 *009	+008	*090090 *090975
22		1.0316	•117	•083	•143	•050	+010	•090	·000080
23		1.0315	•112	•080	•145	•019	•011	.008	•0/0/199
24		1.0311	•120	•084	•148	•051	•011	+010	•000038
25		1.0305	•126	+080	•146	+040	•010	•003	·000074
26	• •	1.0314	•125	*079	•146	•043	·011	•007	+0/10081
$\frac{27}{28}$	• •	1.0311	• 122	•075	• 151	•041	*011	+003	•000086
$\frac{28}{29}$	• •	1.0311 1.0310	•126 •125	•081 •079	•147 •147	•042 •049	*011 *000	*009	+000109 +000079
30	•••	1.0310	•122	.079	•139	•045	+011	+003	•000063
31*		1.0317	•097	+082	•125	•046	•007	•007	•040962
32		1.0304	•117	•077	•138	•0/30	+011	+003	•000084
33		1.0304	·124	.075	•140	•055	•011	•00*)	·000088
34		1.0313	•110	+078	•139	+048	•010	•00)	•000078
35		1.0307	•120	•082	•145	•049	•014	•0.)3	.000000
$\frac{36}{37}$		1.0312	•123 •172	•090	• 150	*045	*007	•093	•090067
38		1.0314	•121	*082 *070	•145 •151	*041 *042	+010	+018	·000080
39*		1.0303	•079	•076	.147	• 045	•011	•003	000092
40		1.0312	•126	•075	•153	•046	·010	•0'0	+000105
41		1.0314	•119	•081	.112	*047	•011	*093	+000096
42		1.0304	•119	•076	·148	+050	*012	+003	•000065
43 44	• •	1.0319	•131	•082	•154	•046	*011	*010	autores t
44 45*	• •	1.0314	*126 *117	•078 •080	•153 •145	*053 *040	*011 *010	+000 +008	•000101 •000068
46	•••	1.0310	•119	+081	148	•050	•011	•000 L	•000367
47		1.0315	•089	•074	•136	•051	.011	+000	·000067
48*		1.0319	+134	•092	•145	•048	.010	•012	·000070
<b>49*</b>		1.0310	•120	•0"8	•148	•047	•010	•003	
50		1.0299	•112	•077	*139	·050	•011		· •
$\frac{51}{52}$	**	1.0306	125	•085	•151	•053	·013	.000	• •
ə∠ 53	••	1 • 0313 1 • 0298	•118 •135	*0_0 *079	•148 •157	•048 •058	•011 •019	*003 *012	• •
54*		1 * 0320	•122	•038	•156	•058	•013	+003	• •
55		1.0323	+111	•031	•152	•046	•011	•007	• •
56		1.0320	•116	•0.50	•150	•051	.012	•00)	
57		1.0301	· 095	*()79	.142	•0 #1	•012	•013	
58		1.0300	*116	+080	•148	•040	·011	•003	
59	••	1.0298	•102	*080	•147	•012	•010	•003	
$\frac{60}{61}$		1.0316	•128	•086	•149	•041	•010	•000	
62	•••	1.0306 1.0313	*120 *123	•081 •078	•148 •151	*010 *042	*011 *011	•003	••
63		1.0316	•125	•0.80	•150	•042	•010	-023	••
64*		1.0318	128	+098	+148	•055	•011	•0:2	
65*		1.0316	•108	•080	•139	*052	•011	•007	
66	•••	1.0324	·118	·089	·142	•049	·012	•00.)	

TABLE I .--- (MINERAL CONSTITUENTS. )

### R. C. Hutchinson:

Num	ber.	Specific Gravity.	Per- centage Ca.	Per- centage P.	Per- centage K.	Per- centage Na.	Per- centage Mg.	Per- centage S.	Per- centag Fe.
37		1:0323	·110	·084	•144	*047	*009	-008	
38*		1.0310	•131	. 081	-152	•046	•011	•011	
39*		1.0316	•114	•035	°148	·047	*01L	+003	
70*		1*0317	• 135	+082	·154	•051	*016	·011	
1*	• •	1.0316	•119	•080	•148	*018	*011	•009	
'2 '3*	••	1.0295	•103	· 030	•146	•044	*010	•009	• •
4*	• •	1.0300	•127 •130	+036 +088	*150 *154	$0.12 \\ 0.18$	*011 *013	*009 *011	
5	•••	1*0330	•111	• 0.30	•148	•049	*011	•007	••
6		1.0317	·118	·082	*145	•050	•011	•007	•••
7		1.0316	·125	+084	·148	•052	•010	•008	
8	• •	1.0333	*130	•051	·151	•051	+011	•010	
9	• •	1.0321	• 125	•080	• 144	+056	•011	•009	• •
$\begin{array}{c} 0 \\ 1 \end{array}$		$1.0328 \\ 1.0320$	*121 *133	*079	•143 •147	•051 •049	•011 •015	·010	• •
12	••	1*0338	•121	• 036	•143	•045	•013	*010 *000	• •
3		1.0316	•119	• 032	138	•051	•010	•009	•••
4		1.0308	•122	•039	+127	•041	•011	+097	•••
5		1.0307	•124	·078	•130	•039	•010	+00S	
6		1.0321	•099	·080	•119	•039	•00.)	•003	
7	• •	1.0315	• 122	•081	*134	+043	•011	•009	• •
8		$1 \cdot 0300$ $1 \cdot 0303$	•123 •127	*084 *036	•130	•039	•011 •010	·006 •004	* *
0	**	1.0310	-185	+038	100		010	*00±	••
ŭ –		1.0302	•129	.083			.012	•003	••
$2^{*}$		1.0308	*120	+083	+140	*053	·011	•007	
3*		L•0319	• 124	.038	152	•055	•012	*008	
4	• •	1.0308	•114	•030	•138	*054	•012	•007	
15* 16		$1 \cdot 0319$ $1 \cdot 0320$	•119 •130	*039 *036	•136 •143	*015 *048	+011 +011	*007	• •
07		1.0315	•121	+0.85	143	•045	*009	•010 •011	• •
8*		1.0313	*094	•074	140	•011	•013	•003	••
9*		1.0311	•123	·031	•149	*052	.008	•010	
00		1.0317	•124	+087	• 156	+048	•011	•000	
)1		1.0318	•129	·088	•150	*050	·012	•000	
)2	• •	1.0319	•127	•034	*146 *132	•051	*010	*003	• •
)3 )4	• •	1.0321 1.0312	·105 ·122	*034 *074	•132	*017 *042	•012 •011	*007 *003	• •
)5	•••	1.0309	•121	•078	+143	052	•010	*003	• •
)š		1.0308	•120	+032	• 146	·0 18	•011	•000	
)7		1.0317	•132	•078	<ul> <li>156</li> </ul>	*046	.012	•000	
13		1.0317	•120	.080	·153	•044	+011	•003	
10	• •	1.0312	•124	°079	•146	*045	•003	•095	
0	• •	$1 \cdot 0315$ $1 \cdot 0314$	*129 *120	*079 *032	*150 *138	•048 •055	*012 *011	*008	• •
2		1+0313	•114	•081	•142	•030	•013	*007 *007	• •
3		1.0319	• 120	.072	•140	•048	.010	•012	• •
.4		1.0328	*135	·088	·154	•045	•013	•010	
5		1.0319	• 124	•079	•134	•050	•011	•00.)	
6		1.0318	*134 *129	+083 +085	+138	•019	•011	•093	
7	• •	1.0314 1.0315	•129	*082	•140 •128	*050 *056	*011 *011	*012	
9		1.0312	132	*032	•145	*052	*011	*003 *011	• •
ő		1.0322	•130	.085	·142	•054	*014	•010	
1		1.0312	•126	•075	•135	•045	·011	.010	
2		1.0313	*124	•087	•138	•045	*013	•011	
3		1.0324	• 127	*0%6	*135	•016	*01L	•000	
4 5		$1 \cdot 0329 \\ 1 \cdot 0318$	+121 +100	*073 *076	•138 •124	*048 *048	*000	•038	• •
13 16	1.1	$1 \cdot 0318$ $1 \cdot 0317$	•116	*076	•124 •130	*048	*010 *010	*003 *007	• •
19 17		1.0314	•117	+080	+148	*040	*010 -	*010	
8		1.0301	· · 112	+070	·145	·012	*003	·010	••
9		1.0311	+122	+084	·150	•048	*012	•000	
30		1:0312	•119	•036	-146	•049	•011	•008	
31		1.0324	+125	*078	•148	*052	•012	•007	
3 <b>2</b> 33		1*0314 = 1*0309	·102 ·121	*066 *085	+138 +141	•041 •050	*013	*003	
33 34	• •	1.0303	•121	+055	* 140	*050	*011 *011	*008	• •
35		1.0303	·119	• 079	143	-046	•010	•008	
36		1.0309	•114	.072	•141	•058	•011	•007	•••
		1.0325	· 129	+077	+153	·056	·013		

## TABLE I.-(MINERAL CONSTITUENTS)-continued.

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Numb	er.	Specific Gravity.	Per- centage Ca.	Per- centage P.	Per- centage K.	Per- centage Na,	Per- centage Mg.	Per- centage S.	Per- centage Fe.
138		1+0329	.121	•084	• 149	•040	·011	·008	
39		1.0319	•105	•075	•147	• 045	·008	.007	
40		1.0329	•114	·086	•148	.050	*012	·008	
41		1.0316	• 094	·078	•119	· 045	.011	.007	
42		1.0313	+118	074	• 134	• 048	·010	•012	
43		1.0318	• 097	078	·145	*050	·008	· 009	
44		1.0311	•110	082	.132	•038	•009	.009	
45		1.0323	·118	•078	•131	*056	•011	•010	• •
46		1.0320	•119	• 082	•137	· 052	*010	·012	
47		1.0317	•108	• 089	•141	054	· 010	· 008	
48		1.0301	•140	·085	•158	•050	*011	· 009	
49		1.0308	•126	•083	•153	050	*011	·003	
50		1.0300	• 123	•082	•151	·042	·012	+007	
51		1.0326	•125	•089	•147	•046	•011	.008	• •
52		1.0323	•117	•084	•148	+050	•012	•007	• •
53		1.0326	•110	•080	•144	·048	•011	·010	
54		1.0318	•095	•071	•135	*049	-010	·010	
55		1.0313	•112	•078	· 132	•012	•011	·009	
56		1.0310	.121	•081	•134	•048	·011	·010	
57		1.0309	•131	•087	•147	.043	·011	·011	
58		1.0317	•133	.078	.146	.050	•014	.012	
59		1.0312	•121	.078	•142	*053	·010	•009	
60		1.0321	•125	•080	•150	• 044	·011	·008	
61		1.0315	•114	·076	•141	.046	·011	·009	
62		1.0316	•126	·079	•138	-048	•012	•010	
63		1.0315	·128	.078	•154	.052	.011	•009	
64	11	1.0315	·119	·089	· 150	•054		•008	
65		1.0311	·135	•079	•157	+056	·011	•014	
66		1.0318	•116	·078	•148	+048	•011	+009	
67		1.0311	•129	*082	•145	* • 046	·011	·010	
68		1.0316	• 120	·081	+139	·048	·011	•009	
Mean		1.0318†	•120	•082	•144	·048	·012	·009	.000087

TABLE I .--- (MINERAL CONSTITUENTS)-continued.

† See text.

TABLE I.--- (VITAMIN CONSTITUENTS.)

	4:37 4:53 4:55	*00082 *00036	51			
· · · · ·	4.53			19	•22	.00260
			48	25	•17	•00226
		*00070	48	17	•12	·00020
	4.95	· 00048	54	20	·22	*00034
	4.14	•00060	53	24	•17	*0001#
	3.95	•00105	35	22	.16	*00036
	4.75	•00059	33	20	•18	·00074
						*00186
						*00046
						100200
						100048
						*00180
						•00250
						00178
						•00079
						•00087
						+00128
						*00230
						·00198
						•00290
						·00052
						·00248
						•00032
1						•00094
						•00224
						·00108
	· · · · · · · · · · · · · · · · · · ·	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$

### R. C. Hutchinson:

Number.	Percentage Fat.	Percentage Carotene in B.F.	I.U. Vitamin A /gm. B.F.	I.U. Vitamin B <sub>1</sub> /100 ccs.	Mgms. Vitamin B <sub>2</sub> /100 ccs.	Percentage Vitamin C.
27	4.10	·00075	66	15	· 22	·00056
28	5.22	*00062 .	61	22	•16	00224
29	4.65	•00059	29	18 19	•15 •14	*00240 *00054
30	$4 \cdot 48 \\ 4 \cdot 05$	·00046 ·00031	31 58	19	-18	·00032
$\begin{array}{ccc} 31 & \cdots \\ 32 & \cdots \end{array}$	5.45	•00112	67	19	•17	*00044
33	5.10	·00107	62	17	·16	·00034
34	5.02	·00113	46	20	•17	•00034
35	5.02	·00155	61	19	•15 •19	*00050
36 37	4 · 23 6 · 10	+00036 +00204	53 38	19 18	•18	+00032 +00048
37    38	5.10	*00118	45	25	•16	00258
39	4.58	·00075	38	20	·18	.00303
40	3 • 92	·00084	45	20	•12	·00240
41	4.67	*00112	56 55	$     24 \\     18   $	•15 •19	$00172 \\ 00192$
42     43	4.30	·00058 ·00084	66	19	*16	•00206
$43   \\ 44  $	4.47	+00078	50	22	•21	•00192
45	5.52	·00153	60	15	•19	·00048
46	5 • 30	*00062	36	18	•21	*00254
47	4 • 20 4 • 50	+00070 +00093	40     40     40	15 21	•16 •18	*00224 *00192
48 49	4.15	+00103	45	14	•17	•00234
50	4.00	·00142	33	14	·21	·00172
51	5.12	+00094	37	15	•16	·00238
52	4.55	•00090	34	18	·21 ·12	*00168 *00188
53	$4 \cdot 18 \\ 3 \cdot 98$	+00189 +00088	41 56	17     19	• 21	*00188
$54 \dots 55 \dots$	4.45	•00133	57	17	·26	*00220
56	3.85	·00145	68	19	+18	*00160
57	4 • 93	·00148	74	17	·22	·00046
58	4.65	·00078	52 54	$     19 \\     15   $	*17 *16	·00192 ·00101
59 60	$4 \cdot 75 \\ 4 \cdot 90$	*00115 *00138	63	19	+25	+00186
60 61	5.62	•00165	70	17	•21	·00238
62	3.65	+00109	60	18	•17	·00182
63	3.50	*00158	62	20	· 12 · 22	·00196
$\begin{array}{ccc} 64 &\\ 65 &\end{array}$	4 • 05 3 • 95	+00045 +00150	59 66	19 18	•24	00192 00190
65 · · ·	5.00	•00116	56	15	•12	·00224
67	4.50	·00148	51	18	.18	·00122
68	4 · 70 5 · 15	·00148	55	19	•19	·00184
69	5 • 15 4 • 90	*00206 *00228	72	15     18	*22	$+00252 \\ +00182$
$\begin{array}{ccc} 70 & \cdots \\ 71 & \cdots \end{array}$	5.73	•00228	38	25	.17	•00184
72	4.80	·00190	52	21	•18	·00162
73	6.75	·00119	55	20	•19	*00240
74	6.00 3.80	•00228 •00107	40	$\frac{14}{20}$	*22 *17	*00123 *00250
$\begin{array}{ccc} 75 & \ldots \\ 76 & \ldots \end{array}$	4.55	· 00125	45	24	+12	*00222
77	3.60	·00119	36	18	•18	$\cdot 00062$
78	3.43	·00141	49	14	18	·00108
79	3.15	*00138	39 35	15     18	·13 •18	*00188 *00092
	$\frac{4 \cdot 00}{3 \cdot 40}$	·00145 ·00174	67	18	•20	·00156
	5.30	00242	45	17	•19	·00120
83	3.83	·00135	38	19	• 17	·00098
84	3.95	•00070	33	17	•12	00250
85 86	3·80 3·45	*00183 *00098	37	20 21	·12 ·14	$+00222 \\ +00148$
86 87	4.25	•00035	47	19	•20	·00260
88	3 • 85	+00111	29	17	•09	·00306
89	3.65	·00167	34	19	•17	*00150
90	3 · 70 3 · 90	·00060 ·00245	35 33	19     20	·18 ·27	*00054 *00044
91 92	3.00	+00245	46		•17	•00096
92 93	4.60	*00078	46	24	17	*00084
94	4 • 40	*00081	36	19	•18	·00093
95	5.15	*00101	45	16 19	$^{+17}_{-20}$	·00139 ·00200
96 97		·00098 ·00105	47 68	19	$^{+20}_{-19}$	*00200
97 98	4.12	*00065	52	20	·16	$\cdot 00192$

TABLE I.- (VITAMIN CONSTITUENTS)-continued.

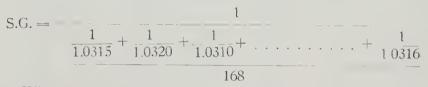
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Nur	nber.	Percentage Fat.	Percentage Carotene in B.F.	I.U. Vitamin A /gm. B.F.	I.U. Vitamin $B_1$ /100 ccs.	$\begin{array}{c} \text{Mgms.}\\ \text{Vitamin } \text{B}_2\\ /100 \ \text{ccs.} \end{array}$	Percentage Vitamin C.
99		3.98	·00058	44	25	*13	·00128
100	••	4.33	•00019 •00155	29	20	• 16,	•00130
101 102	••	4.33 5.34	*00155 *00134	$\frac{56}{44}$	18	•17	*00084
103	••	4.88	*00221	54 54	$\frac{24}{19}$	•14 •16	*00238 *00156
104		4.28	•00190	57	23	*20	•00044
105		4.27	*00105	33	18	•12	*00036
$106 \\ 107$	• •	4 • 63 4 • 35	*00045 *00056	34 35	15	•19	*00172
108	••	4.85	•00005	28	$\frac{28}{20}$	•16 •13	*00068 *00034
109		4.04	•00050	36	18	+17	*00036
110	• •	3.85	*00058	36	18	•15	•00206
$111 \\ 112$	••	4 ° 65 5 ° 25	•00030 •00161	$\frac{40}{68}$	15	*19 *30	•09044
113	••	3.10	*00172	61	19     21	• 12	*00042 *00302
114		4.15	*00121	56	14	•15	•00117
115	••	4.05	·00150	60	14	•15	·00192
$116 \\ 117$	•••	5 • 45 5 • 10	*00065 *00084	$\frac{40}{37}$	19	•16 •10	·00151
118	•••	5.05	•00120	37 33	15     18	*18 *17	•00192
119		4.05	·00114	38	17	•17	*00166
120	• •	4.33	*00074 *00036	45	26	*17	·00986
121 122	•••	$6 \cdot 10 \\ 5 \cdot 00$	+00036 +00018	67 35	$\frac{14}{23}$	20 16	*00268 *00192
123	••	4.68	•00056	38	19	*18	.00258
124		4.30	•00098	57	20	•28	•00136
$125 \\ 126$	• •	4.00	*00145	62	22	•17	·00248
120	•••	$\frac{4*90}{5*63}$	•00200 •00068	$\frac{46}{61}$	23 13	*18 *15	·00220
128		4.90	+00064	35	24	•15	*00068 *00232
129 -		6 • 45	+00079	39	25	• 22	·00240
130	• •	6.10	*00082	49	27	*20	·00168
$131 \\ 132$		$3 \cdot 70 \\ 4 \cdot 65$	•00067 •00104	$\frac{45}{36}$	22 18	•16 •16	$00210 \\ 00254$
133		3.50	+09092	53	18	•15	•00224
134		3.53	·00040	38	20	•14	·00322
l 35 136	• •	$3 \cdot 05 \\ 4 \cdot 10$	*09034	45	21	•17	
130		3.30	*00112 *00034	$\frac{28}{45}$	$\frac{17}{16}$	*19 *18	••
138		5.20	•00058	56	15	•23	
139		3.73	·00065	55	23	•16	
140 141		3.85	*00153 *09078	66	25	• 15	
142		4.25	+00103	33 40	$15 \\ 22$	*17 *17	• •
143		3.75	•00093	35	19	•29	
144	• •	3.45	•00070	52	20	•22	
145 146	• •	$3 \cdot 70 \\ 3 \cdot 95$	*00990 *09102	38 44	21	•23	••
147		5.15	·00132	62	$     14 \\     16   $	*15 *22	••
148		4.12	·00145	55	20	•17	
149 150	• •	3.98	·00148	51	· 22	•18	
150 151		$4 \cdot 33 \\ 5 \cdot 34$	*00165 *00109	56 50	18 24	*15 *19	• •
52		4.88	+00058	60	24 22	• 19	
153		4.28	•00970	36	15	•19 j	
154 155		4 · 27 4 · 63	*00107	40	18	•17	• •
199 156		4*03	*00036 *00075	$\frac{30}{45}$	15	·17 ·18	••
57		5.10	·00084	56	18 19	•22	••
58		4.20	·00078	66	25	*17	
l59 l60	• •	4.75	•00068	59	15	*14	
61		$3 \cdot 90 \\ 4 \cdot 10$	*00070 *00020	62 60	11 18	•19 •14	••
162		4.00	+00135	60	$\frac{18}{20}$	*10	• •
63		4.65	•00111	63	25	•18	
l64 l65	• •	5.20	*00167	54	21	•19	
100 166	••	$4 \cdot 85 \\ 3 \cdot 90$	*00063 *00065	$\frac{62}{51}$	20	•19	••
67		4.20	*00058	65 1	$     14 \\     19 $	•13 •18	
68		4.06	•00079	43	20	•17	
Mea	n	4.51	·00104	47.6	18.8	•17	•00156

TABLE I .-- (VITAMIN CONSTITUENTS)-continued.

#### General Discussion of Results.

In Table I, the mean specific gravity was not obtained by summing all the specific gravities and dividing by their number. To average specific gravities they should be first calculated to specific volumes, these averaged, and the average specific gravity deduced from the average specific volume. The average specific gravity then was calculated from the following:—



When the specific gravity and fat content of milk are known, the total solids and solids-not-fat may be readily calculated by means of Richmond's (1930) milk scale. The mean total solids and solids-not-fat content of the milk samples estimated in this manner are 13.50 and 8.99 per cent. respectively.

A slight direct correlation was found to exist between the solids-not-fat and phosphorus content of the samples, the Pear-sonian coefficient of correlation being .40.

In Table III, the composition of milk from different breeds of cattle including the Australian Illawarra Shorthorn is compared. It will be seen that the composition of milk from Australian Illawarra Shorthorn herds resembles that from Ayrshire rather than Guernsey or Jersey herds. All samples were from afternoon milkings.

		Total §	Solids Pe	r Cent.	Fa	l Per C	ent.		ids–not- Per Cent	
Breed.	Number of Samples.	Maximmı.	Mnnmun.	Avelaut.	Maximim.	M. trituuto,	Average	Maxumum.	Minimum.	Ателице
Australian Illawarra		%	%	%	%	%	%	0/ /0	%	%
Shorthorn Ayrshire Guernsey Jersey	$\begin{array}{c}16\\12\\4\\10\end{array}$	$\begin{array}{c} 14\cdot 32 \\ 14\cdot 10 \\ 14\cdot 81 \\ 15\cdot 53 \end{array}$	$\begin{array}{c} 11^{+}67 \\ 11^{+}94 \\ 14^{+}36 \\ 13^{+}04 \end{array}$	$\begin{array}{c} 13 \cdot 02 \\ 13 \cdot 18 \\ 14 \cdot 56 \\ 14 \cdot 25 \end{array}$	$4^{\circ}93 \\ 5^{\circ}05 \\ 5^{\circ}45 \\ 6^{\circ}10$	$3 \cdot 35 \\ 3 \cdot 50 \\ 5 \cdot 30 \\ 4 \cdot 50$	$     \begin{array}{r}       4 \cdot 35 \\       4 \cdot 43 \\       5 \cdot 37 \\       5 \cdot 36     \end{array} $	9.39 9.05 9.36 9.43	8.32 8.44 9.06 8.54	8.67 8.75 9.19 8.89

TABLE III.-COMPOSITION OF MILK FROM COWS OF DIFFERENT BREEDS.

VITAMIN A AND CAROTENE.

Very fcw figures have been published on the vitamin A content of butter-fat which are comparable with those obtained during this investigation. In the large majority of cases the estimations have been carried out on butter-fat from a small number of cows, a particular breed of cow, or on butter-fat obtained during a

#### TABLE II .- FREQUENCY DISTRIBUTION OF RESULTS.

																				Mean	-																-					
Specific D. Gravity F.	295 1		297 0	298 2	299 1	300 4		30 <u>2</u> 1	303 1	304 4	305 2	306 3	307 2	<b>30</b> 8 6	309 6	310 11	311 11	312 9	313 8		315 12	$\begin{array}{c} 316\\11\end{array}$	$317 \\ 11$	318 7	319 8	320 8	321 5	322 1	323 4	324 3		326 2	327 0	328 2	329 3	330 2	331 1	332 2	333 1	•••	••	•••
Fat D. F.	•••				•••	3·1 2	3·2 2	3·3 1	3•4 3	3·5 5	3.6 1	3°7 6	3·8 4	3.9	4.0 13	4·1 11	4·2 8	4·3 13	4•4 6	4°5 6	4.6 8	<b>4·7</b> 8	4·8 5	4*9 9	5°0 5	5·1 11	5°2 8	5.3	5•4 2	5°5 3	5·6 4	5·7 1	5.8 0	$5 \cdot 9 \\ 0$	6°0 1	6°1 3	6*2 1	6·3 0		$6.5 \\ 1$		
Carotene D F.				•••	 	··· ··		30 3	36 2	42 1	48 6	54 3	60 16	66 11	72 8	78 12	84 12	90 6	96 9			114 9	120 6	126 2	132 3	13∢ 6	144 6	150 6	156 5	162 1		174 2	180 0	186     1		198 1	-04 2	210 0		222 1	228 2	< 234 3
Vitamin A D. F.		:: .		•••			•••	•••		28 3	30 5	32 1	34 13	36 17	38 11	40 12	42 2	44 6		48 4	50 5		54 9	56 14	58 5	60 8	62 10	64 3	66 6	68 7	70 1	72 1	74 1	::			•••	••			•••	 
Vitamin B <sub>1</sub> D. F.											10 2	11 1	12 0	13 1	14 12	15 18	16 4	17 12	18 25	19 28	20 25	21 8	22 8	23 4	24 8	25 8	26 1		28 1					::				•••	••	•••	•••	···
Vitamin B <sub>s</sub> D. F.												9 1	10 1	11 0	12 14	13 4	14 9	15 14	16 20	17 31	18 21	19 18	20 8	21 6	22 11	23 3	· 24 · 1	25 1	26 1	27 1	28 1	29 1	30 1	 	 			- 1	•••	••	•••	<u>-</u>
Vitamin C D. F.	•••		••	•••	36 10	44 8	52 7	60 2	68 2	76 2	84 5	92 3	100 4	108 2	116 1	124 3	132 3	140 2	148 3		164 3	172 5	180 4	188 8	196 11	204 4	212 1	220 4	228 7	236 5	244 4	252 8	260 5	268 1	$276 \\ 0$	284 0	292 1	300 2	308 1	316 0	324 1	•••
Caicium D. F.			103 0			106 0	107 1	108 3	109 1	110 4	111 3	112 5	113 0	114 7	115 0	116 6	117 5	118 7	119 11	120 10	121 10	122 8	123 5	124 7	125 12	126 8	127 5	128 4	129 8	130 7	131 3	132 2	133 2	134 3		136 0		138 0	139 0		••	• •
Phosphorus D F.		• •	•••	66 1	67 0	68 0		70 1	71 1	7 <u>2</u> 1	73 1	74 5	75 6	76 4		78 17	79 14	80 23	81 12	82 16	83 5	84 11	85 8		87 6	88 8	89 6	90 2	91 1		93 0		95 0	96 0	97 1						•••	•••
Potassium D. F.	<126 4	126 0	127 1	128 1		130 3	131 1		133 0	134 5	135 4		137 I	138 11	139 5	140 8		142 9				146 9	147 9	148 21	149 3	150 12	151 6	152 4	153 5	154 5	155 0	156 4	157 2	158 1						•••	•••	••
Sodium D. F.		• •	•••	'						38 1	39 4	40 6	41 7	42 13	43 2	44 5	45 15	46 12	47 7	48 22	49 10		51 10	52 8	53 5	54 5	55 5	56 4	57 1	58 3	59 1	60 1	::	::						· · ·	•••	••
Magnesium D. F.		••		 											07 2	08 3	09 13	10 28	11 75	12 24	13 13	14 5	15 1	16 1	17 0	18 0	19 1			::				::	 		 			··· ··		···
Sulphur D F.			::	•••	•••										04 1	05 0	06 4	07 20			10 24		12 9	13 1	14 1	15 0	16 1		•••	::			::			::			•••		•••	• •
Iron D. F.	• •	•••	•••	• • •	•••				•••						060	005 6	070 5	075	080	085 3	090 7	095 1	100 2	105 2	110 4	115 0	120 0	125 1	< 125	::		::	::		•••			•••			•••	••

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particular season. Such results, obtained by spectrophotometric methods, have been published by Baumann et al. (1934), Becson (1935), Booth et al. (1934), Gillam et al. (1936), Peterson et al. (1935), and Sutton and Kraus (1936). Sherman and Sherman (1937) give the mean vitamin A value of 86 samples of butter-fat as  $50.60 \pm 1.80$  international units per gram, which differs only by three international units from the figure in Table I. Crawford et al. (1932) have determined the vitamin A activity of Australian butter-fat biologically, but because of the method used, the results are not directly comparable with those in this work. The frequency distribution table shows carotene to be a very variable constituent.

#### VITAMIN $B_1$ .

Very few figures are to be found for the vitamin  $B_1$  content of milk. Baker and Wright (1935) have published 23 international units per 100 ccs., which is four international units higher than the results obtained during this investigation, whilst Pyke (1937) found two samples contained 6 and 11 international units per 100 ml.

#### VITAMIN B<sub>2</sub>.

Comparatively little work has been published on the vitamin  $B_2$  content of milk. The figures .2 to .3 mgm. per 100 ccs., .1 mgm. per 100 ccs., and .176 to .26 mgm. per 100 ccs. have been obtained by Euler and Adler (1934), Kuhu ct al. (1934), and Whitnah et al. (1937) respectively.

#### VITAMIN C.

The vitamin C content of milk has been investigated by many workers and appears to vary from < .3 (Levy and Fox, 1935) to 2.92 (Whitnah and Riddell, 1937) mgms. per 100 gms. 1.77 mgms. per 100 gms. which is slightly higher than the figure (1.56 mgms. per 100 gms.) given in Table I., is the mean of nineteen results obtained by Levy and Fox (1935), Whitnah and Riddell (1937), Ranganathan (1935), Fujita and Ebihara (1937). Meulemans and De Haas (1937). Rasmussen et al. (1936), Harris and Ray (1935), Correns (1937), Birch et al. (1933), Whitnah and Riddell (1936), Van Wijngaarden (1934), Kon and Watron (1937), Riddell et al. (1936), and Ferdinand (1936).

#### POTASSIUM, SODIUM, MAGNESIUM, AND SULPHUR.

Sherman (1936) gives .143 for the percentage of potassium in whole milk, Richmond (1930) .150, Crichton (1930) .168, and Trunz (1903) gives .136 and .149 per cent. For sodium Sherman gives .051 per cent., Richmond .037, Crichton .056, and Trunz gives .032 and .042 per cent. For the amount of magnesium in whole milk, Forbes et al. (1917) and Hart et al. (1909) each give .011 per cent., Sherman .012 per cent., Forbes et al. (1918) and Richmond .013 per cent., while Trunz gives the range .011 to .015 per cent. and .012 to .017 per cent. König (1904) gives .007 per cent. as the amount of sulphur in whole milk which has been ashed. (Approximately 72 per cent. of the sulphur in milk is lost in ashing). These figures are in good agreement with those obtained during this survey.

#### IRON.

Table IV. gives the iron content of milk according to various investigators :---

Investigator.	Parts per Million.	Investigator.	Parts per Million.
Davies (1931) Edelstein and Csonka (1912) . Elvehjem (1926) Fendler et al. (1914) . Friedjung (1901)	$\begin{array}{c} \cdot 62 - 84 \\ \cdot 1^* 5 - 2^* 4 \\ \cdot 28 - 49 \\ \cdot 35 - 36 \\ \cdot 2^* 8 - 8^* 4 \\ \cdot 84 - 1^* 82 \\ \cdot 35 - 4^* 69 \end{array}$	Langstein (1911) Macfarlane (1932) Nottbohm and Dorr (1914) Sherman (1936) Soxhlet (1912) Trunz (1903) This (recettantian	$\begin{array}{cccc} & & & & & & & \\ & & & & & & \\ & & & & $
			l

TABLE IV IRON CONTENT OF MIL	TABLE	1VIRON	CONTENT	OF	MILK
------------------------------	-------	--------	---------	----	------

These figures vary considerably, but the figure for Australian milk is well within the common range.

#### CALCIUM.

Table V. shows that the mean figure for the calcium content of Australian milk agreees very well with figures obtained by overseas investigators.

Inv	estigat	or.		Percentage Ca.	Percentage P
Bunge (1901)			 		• 077-• 084
Decem 1 XX7244 (1095)			 	· 083-·141	·071-·117
CI 0.11			 	· 132	· 102
Childen (1020)			 	$\cdot 119$	· 100
Davies and Provan (19			 	· 124	· 105
TA . T	• •		 	· 103	·078
17 1. 1. 1 (1010)			 	· 117	* 094
California at al (1099)			 	• 128	· 107
(0001) h. A. T			 	·084-·097	.076
Hutchison (1906)			 		.096
1° uto monus (1000)			 	• 119	·094
P-4 (1000)			 	127	· 098
LT TH A THORAGE (			 	· 112	· 080
Maine at al (1096)			 	· 100	· 087
Meigs et al. (1926)			 	· 132	• 114
Distance of (1000)			 	· 109	. 096
Sheehy (1921)			 		· 090
Sherman (1936)			 	· 118	· 093
Sommer and Hart (192	6)		 	135	• 095
Sommer and Hart (192			 	.124	· 113
Sommer and Hart (192			 	• 142	·102
Sommer and Hart (192			 	129	* 103
Trunz (1903)			 	137	· 083-· 103
Trunz (1903)	•••		 I	• 128	· 089-·100
Wellmann (1937)			 	· 118-·146	·096-·155
This Investigation	• •		 	· 120	· 082

TABLE V.- CALCIUM AND PHOSPHORUS CONTENT OF MILK.

PHOSPHORUS.

Table V. compares the mean figure for the phosphorus content of Australian milk with twenty-five figures obtained by overseas investigators. It will be seen that three of the overseas figures are slightly lower and one practically equal to the Australian figure, whilst twenty-one are definitely higher. Hence it was considered reasonable to conclude that the phosphorus content of the Australian samples was low.

However, it was within the normal range according to Bunge, Burr and Witt, Wellman (see Table V.) and the following investigators. Sommer (1929) gives the range .068 to .119 per cent., Cranfield et al. (1927) (673 samples) .076 to .135 per cent., although 80 per cent. of the samples fell between the range .092 to .111 per cent., and Crichton (1930) has found it to be .073 to .127 per cent. for 220 samples.

It was at first thought that this low figure may be due to the method of estimation, but this was checked against the wellestablished method of titrating the ash solution with uranium acetate using potassium ferrocyanide as external indicator when both methods gave results in good agreement.

It has been recognized for some decades that many Australian soils, particularly the superficial soils, are deficient in phosphorus and this has been indirectly proved by the remarkable results which follow the use in New South Wales, Victoria, and Tasmania of phosphatic fertilizers. In an article by Cherry (1907) Victorian clay soils were compared with American clay soils, the clay soils being chosen because they contained a higher percentage of phosphorus than other soils. It was shown that Victorian clay soils contained 63 parts of phosphoric acid per 100,000 parts, whereas American clay soils contained 207 parts per 100,000. The Mallee soils were even lower, containing only 47 parts per 100,000.

During a survey conducted by Henry and Benjamin (1933), it was found that of 56 soils analysed from the southern eoastal belt of New South Wales, only eight samples contained as much as five parts per 100,000 of citrate soluble phosphoric acid, the remaining 48 samples averaging 2.3 parts per 100,000.

Many other papers have been written containing data on the low phosphorus content of Australian soils. In the eastern States there is the work of Taylor and Hooper (1938), Taylor and England (1929), Taylor and Penman (1930), and Taylor et al. (1933), whilst in Western Australia there is the work of Hosking and Burvill (1938) and Teakle (1929).

In unpublished work carried out in this laboratory it was found, after analysing eighty-six samples of soil collected from dairy farms situated on the northern coastal belt of New South Wales, that each of these soils was very low in phosphorus. Many research workers have found that the mineral content of pastures, fodder crops, cereals, &c., is related to the mineral content of the soils on which they are grown. In order to investigate this, one hundred and fifty-six samples of pastures and fodder crops were collected from the eighty-six previously mentioned dairy farms at the time the milk and soil samples were collected. The phosphorus content of these samples, dried at 100°C., was determined by a method similar to that used for milk, and the results together with comparative figures by overseas investigators are given in Table VI.

Feed.		Remarks.	P Per Cent.	Number of Samples,	Country.	Authority.	
Mixed	pasture	From fertilized	· 149	11	Australia	This investigation	
55	,,	From unimproved land	·178	46	,, .	*3 59	
> 9	*7	Good pasture	• 46	,	New Zealand	Rigg and Askew (1929)	
• •	,,	Poor hill pasture	• 33		53 53	33 17 39	
13	* *	From both man- nred and un- manured land	• 335	40	England	Armstrong (1907)	
7 °	22	From cultivated	· 32	24	England and Wales	Godden (1926)	
"	51	From hill country	•26	35	Scotland	a 33 35	
,,	33	From natural pas- ture	· 29	22	,,	Orr (1929)	
3.3	,,	From cultivated	· 32	24	,,	5.5 5.5	
,,	۰,	From poor hill country	· 26	35	,,	* *3 33	
,,	3.2	··· ·· ··	· 29	12	Kenya	Orr and Holm (1931)	
2.9	3 3	· · · ·	· 223	60	Mauritius	Lincoln (1937)	
9.5	1 1	Natural Pasture	.303	86	United States	Newlander et al (1933)	
3 3	> 3	ss ss	•19	96	33 31	Archibald and Benuett (1933)	
Fresh	lucerne*	From fertilized land	• 419	19	Australia	This investigation	
,,	*	From unimproved land	• 411	33	,, ···	93 33	
19	* **	••••••	•625		England	Woodman (1934)	
7.9	>>		• 457		South Africa	Fox and Wilson ()	
Green	oats	From fertilized	· 520	4	Australia	This investigation	
13	,, ·-	From unimproved land	• 554	4	,,	33 21	
Fresh	sorghum	From fertilized	• 097	16	Australia	This investigation	
,,	37	From unimproved land	· 089	5	,, · ·	33 35	
Bran			• 98	5	Australia	This investigation	
2.3			1.32		America	Morrison (1936)	
a ""			1.20		Indiana	Purdue Uni. Agr. Exp.	
,,			1.452	• •	Europe	Sta. (1938) Heubner and Reeb (1908)	

TABLE VI .- PHOSPHORUS CONTENT OF FEEDS.

\* Pre-budding period.

From this table it was seen that the phosphorus content of these pastures and fodder crops was very low. In some cases it was lower in crops grown on fertilized land than in crops grown on unfertilized land. This throws doubt on the testimonies of the farmers who supplied the information regarding fertilization of these lands, although it is possible that soils on which these crops grew may have been so low in phosphorus that the amount of fertilizer used restored but little of the deficiency.

The mean phosphorus content of the mixed milk from these eighty-six dairy farms was .081 per cent. This does not differ appreciably from the mean figure given in Table 1.

The question now arises as to whether the amount of phosphorus ingested by the cow determines the phosphorus content of milk. This question has been investigated by many workers and an excellent review of literature on the subject is included in a paper by Crichton (1930). On summing up the results of the various investigations a conclusion similar to that expressed by Forbes and Keith (1914) developed, namely, that the character of the feed may vary the phosphorus content of the milk, but only within normal limits.

Many of the milk samples collected during this survey were from herds which have been feeding for cow-generations on low phosphorus diets and often in localities where the conditions were so severe that cases of osteophagia were frequently observed (Henry and Benjamin, 1933). Where such conditions prevail, it seems reasonable to assume that this low phosphorus content of the milk can be due to no other cause than the low phosphorus content of Australian pastures and fodder crops, or indirectly to the low phosphorus content of Australian soils. This conclusion does not support that of Kincaid (1911), who analysed four milk samples from Victoria.

#### Correlations between the Vitamin and Mineral Constituents.

In a preliminary examination points were plotted for every value of each of the vitamius A,  $B_1$ ,  $B_2$ , and C against the values for the mineral elements from the same milk sample. Twentyeight graphs were thus obtained. These graphs were carefully examined, but in no instance could any correlation be detected between any two sets of results. Indeed, there was such a lack of correlation that it was considered unnecessary to apply any statistical treatment.

No correlation was found to exist between any of the vitamin and any of the mineral constituents of milk. In view of the opinion expressed by Richmond (1930) that lecithin, because of its phosphorus content, may be an index of the vitamin content of milk, the lecithin content of thirty samples of milk was estimated by the method of Bordas and Raczkowski (1902) and the results examined as before, but again no correlation was detected. The mean figure for the lecithin content of these samples was .066 per cent.

#### A Correlation between the Phosphorus Content of Pasture and the Vitamin A Content of Milk.

Samples of feed as well as samples of milk were collected from most farms visited during this investigation, and during the examination of results, a most interesting correlation was found to exist between the phosphorus content of the pastures upon which certain herds grazed and the vitamin A content of the butter-fat from these herds.

On certain farms, which were comparatively large holdings, the herds were feeding on pasture alone. These farms (indicated by an asterisk in Table I.) were situated in the coastal dairying districts of New South Wales, the most southern being situated in the Richmond district and the majority of others in the Grafton district. According to information gathered from each farmer, these herds had been feeding in the same paddocks for several cow-generations, the pasture feed had never been supplemented with hand feed of any kind and the paddocks had never been artificially fertilized, so that the mineral and vitamin intake of these herds remained fairly constant throughout the year and from one year to another.

The specimens of pasture and milk were collected in late summer when pastures were poorest, but on each farm there was abundance of green grass on which the cattle could feed and, as will be seen later, the diet contained a large excess of carotene. The pasture samples, which were collected with the assistance of the farmers, were taken from many parts of the field and much care was taken to obtain a truly representative sample. The samples were not cut, but plucked by hand in order to simulate a cow's method of grazing, and herbage not eaten by the cow was not included in the sample. On most farms couch (*Cynodon* sp.) was the predominating grass, and on some farms this was the only pasture grass. Paspalum (*Paspalum* sp.) and various species of clover were also common, but native grasses other than couch constituted a very small portion of any pasture.

#### TABLE VII.

(Phosphorus and protein expressed as percentage of pasture dried at  $100^{\circ}C$ .)

I.U. Vitamin A /gm, of B.F.			Breed of Herd.			Percentage P in Pasture,	Percentage Protein in Pasture.	Carotene in Pasture mgms/kg.	
58			Mixed				· 217	10.12	180
38							· 085	9.04	190
60			Jersey				.171	10.51	240
40	• •		Mixed	- •			· 098	6.72	200
45	• •		5,				* 133	6.56	200
56			3.7				· 202	10.73	190
59	• •		••				+ 186	9.05	180
86			_ 11				• 209	10.62	180
55			Jersey				-187	12.43	200
72	• •		Mixed				325	13.15	210
14	• •		>>	• •			· 101	10.20	250
38							• 116	9.72	190
55	• •	• •	- *2				• 169	9.54	190
10	• •		Jersey				-147	10.89	200
46			Mixed				- 096	9.05	200
16		• •	2.9	• •			•111	10.21	190
15	• •				1.1	• •	-186	12.20	180
52			Australian	Illawarra	Short	horn	132	8+45	200
44	• •		Mixed	+ +			• 124	8.35	180

Table VII. gives the amount of vitamin A per gram of butterfat in the mixed milk from these herds and the percentages of phosphorus, protein, and carotene in the pastures upon which the herds were feeding. It will be seen that a direct correlation exists between the vitamin A content of the butter-fat and the phosphorus content of the pasture, the coefficient of correlation being .97.

The carotene content of the pastures was estimated by a method similar to that given by Bolin and Khalapur (1938) and it will be seen from Table VII, that the poorest pasture contained 180 mgnus, of carotene per kilogram or approximately 288,000 international units of vitamin A per kilogram (Fixsen and Roscoe, 1937-38). According to the work of Fraps et al. (1937) and Guilbert and Hart (1935), cows on these pastures were receiving a large daily excess of vitamin A.

Hence it would appear that the vitamin A content of butter-fat is influenced in some way by the phosphorus as well as the earotene ingested by the cow as suggested by Professor Rosedale. However, there may be other underlying factors such as the stage of growth of the pasture, for it will be seen that there is also a considerable degree of correlation between the protein and phosphorus content of the pastures. Unfortunately, most of the milk samples were from mixed herds, and because of this the table loses some of its value. An interesting piece of confirmatory evidence is given by the following fact, however. The mean vitamin A content of five samples of mixed milk from cows feeding on pasture but receiving in addition a daily ration of bran (a rich source of phosphorus) was two international units higher than the mean figure for the results given in Table VII.

No similar correlation was found to exist between any of the other vitamins and the phosphorus or protein intake of the herd.

#### R. C. Hutchinson:

#### Summary.

Fat, vitamin A, carotene, vitamin  $B_1$ , vitamin  $B_2$ , vitamin C, calcium, phosphorus, potassium, sodium, magnesium, sulphur, and iron have been estimated in 168 samples of mixed milk. Methods and results have been presented. The mean results of all analyses appear in the following table:—

Total solids • 13.50 per cent Phe	agnesium '012 per cent.
Solids-not-fat 8.99 per cent Leo	Ilphur '009 per cent.

By comparing the results with overseas figures it has been shown that the phosphorus content of the Australian milk, although within normal limits, was low. It has also been shown that no correlations exist between the vitamin and mineral constituents of milk. Subject to certain conditions, a correlation was discovered between the vitamin A content of butter-fat and the phosphorus content of the pasture.

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